Visible-Light-Mediated Monoselective Ortho C−H Arylation of 6‑Arylpurine Nucleosides with Diazonium Salts

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ABSTRACT: A combined palladium- and photoredox-catalyzed monoselective arylation of 6-arylpurine nucleosides has been developed by employing purine as a directing group via the photoredox reaction, and many functional groups are well tolerated in this direct C−H arylation condition. Various of functionalized purines (nucleosides) which are potentially of great importance in medicinal chemistry could be obtained under visible light irradiation at room temperature within 4 h.

As the universal structural units in RNA and DNA, purine
bases and nucleosides have displayed unique biological
activities, such as extentation and $HCN²$ antiviral, and activities such as cytostatic,¹ anti-HCV,² antiviral, and antimicrobacterial^{[3](#page-6-0)} activities. In particular, C6 aryl- and heteroaryl purine analogues possess high inhibitory activity against Mycobacterium tuberculosis. [4](#page-6-0) Nowadays, nitrogen-based directed C−H bond arylation has been regarded as an efficient method to construct biaryl compounds which are ubiquitous building blocks in biological and pharmaceutical sciences.^{[5](#page-6-0)} Our group first described purine as the new directing group for Pdcatalyzed Csp^2-H bond arylation ([Scheme 1,](#page-1-0) path (i)), but the reaction could not be regarded as an efficient method because of the high temperature (120 °C) and long reaction time (48− 60 h). Later, the Lakshman's group utilized the identical directing group for the arylation of Csp^2-H bonds under ruthenium catalysis ([Scheme 1](#page-1-0), path (ii)).^{[6](#page-6-0)} Although their condition provides a broader substrate scope (especially deoxyribonucleosides), its practical application in synthesis has been limited due to its lower regioselectivity (mono/ diarylation), long reaction time (24 h), and high temperature (120 °C). Thus, exploring an efficient route to obtain C6 arylated purines (nucleosides) with high regioselectivity under mild condition is challenging and fascinating.

Recently, many efforts have been devoted to direct C−H functionalization using combined transmetal with photoredox catalyst.[7](#page-6-0) The concept of visible-light-mediated direct C−H functionalizations has received significant attention as a promising strategy for the cross coupling reactions including cycloaddition, 8 radical addition, 9 alkylation, 10 arylation, 11 and others.[12](#page-7-0) At first, Sanford and co-workers reported liganddirected C−H arylation reactions by using aryl diazonium salts and diaryliodonium salts as cross partners under visible light irradiation, which provided a new strategy for the construction of biaryl compounds [\(Scheme 1](#page-1-0)).^{7a,b} Different directing groups (such as amides, pyrazoles, pyrimidines, and oxime ethers) were examined. However, no example of photocatalyzed purine modifucation has been reported up to now. Based on our longstanding goals on the selective modification of nucleoside analogues and the established work on the arylation reaction, 13 we describe a mild and highly regioselective method for the direct arylation of 6-aryl-purines (nucleosides) using photoredox catalyst, palladium, and diazonium salts under visible light irradiation.

Initially, we began our study by using 6-phenyl-9-benzylpurine 1a and phenyldiazonium tetrafluoroborate 2a as model substrates to optimize the reaction conditions ([Table 1](#page-2-0)). To find if a photoredox catalyst can be used to enable a $Pd(IV)/$ $Pd(II)$ catalytic cycle,^{[7a](#page-6-0)} the dual palladium/photoredox catalytic system was first used to test the arylation reaction. As shown in [Table 1,](#page-2-0) white LED irradiation of a mixture of 6-aryl-9-benzylpurine, aryl diazonium salt, $Ru(bpy)_{3}Cl_{2}.6H_{2}O$, and $Pd(OAc)_{2}$ in MeOH at room temperature under air provided 3aa as a product in trace amounts ([Table 1,](#page-2-0) entry 1). When the reaction was carried out in an inert atmosphere and irradiated with a blue LED bulb, the yield of product 3aa was obviously increased ([Table 1,](#page-2-0) entries 1−3). Subsequently, other palladium sources and transition-mental-catalysts were also screened, which revealed $Pd(OAc)_2$ to be the suitable catalyst for this reaction [\(Table 1,](#page-2-0) entries 4−7).

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Scheme 1. Strategies for the Ortho Arylation of 6-Arylpurine (Nucleoside)

Sanford's work: Drect C-H arylation of arenes by dual palladium and photoredox catalyst

 $Ru(bpy)_{3}Cl_{2}·6H_{2}O$ was the best photoredox catalyst that gave the optimal reaction effect [\(Table 1](#page-2-0), entries 3, 8−10). Control reactions confirmed that both $Pd(OAc)_2$ and Ru- (bpy) ₃Cl₂·6H₂O and light played a crucial role for significant conversion to the product ([Table 1](#page-2-0), entries 11−13). Thus, the optimal reaction conditions were 5 mol % $Pd(OAc)_{2}$ and 2.5 mol % $Ru(bpy)_{3}Cl_{2}·6H_{2}O$ in MeOH at room temperature under N_2 irradiation with a blue LED bulbs [\(Table 1,](#page-2-0) entry 3).

Under the optimized reaction conditions ([Table 1](#page-2-0), entry 3), the substrate scope of C−H arylation reaction was examined. As shown in [Scheme 2,](#page-2-0) the reaction of N9-substituted substrates (including alkyl, benzyl, ester, sugar) with Pd- $(OAc)_{2}/Ru(bpy)_{3}Cl_{2}.6H_{2}O$ catalytic system afforded the corresponding arylation products in 63%−86% yields ([Scheme](#page-2-0) [2](#page-2-0)). The N9 alkyl- or cycloalkyl-substituted substrates reacted smoothly and gave the desired products in higher yields [\(Scheme 2](#page-2-0), 3ba−3da), while substrates bearing some active groups such as bromine, ester, or sugar resulted in slightly lower yields [\(Scheme 2](#page-2-0), 3ea−3ga). It is worth mentioning that the reaction time of the substrate with ester and sugar should be shortened to 2 h in order to avoid generating excessive byproducts (not the diarylation products), and it was necessary to increase the amount of aryl diazonium for their conversion [\(Scheme 2,](#page-2-0) 3fa−3ga).

Subsequently, the tolerance of this reaction to a range of aryldiazonium salts and the electronic effect on the 6 arylpurines were examined [\(Scheme 3\)](#page-2-0). First, a purine nucleoside was reacted with various substituted aryldiazonium salts, and the corresponding biaryl-purine nucleosides were obtained in moderate to good yields (see [Supporting](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.7b00659/suppl_file/jo7b00659_si_001.pdf) [Information](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.7b00659/suppl_file/jo7b00659_si_001.pdf) for competition reactions). The reaction proceeded smoothly with diazonium salts having bromine substituted at each position of the aryl ring, giving acceptable yields of desired products [\(Scheme 3](#page-2-0), 3ge−3gg), except that the yield of the desired product 3gf was somewhat low. This might be due to the steric hindrance of o-Br. Both electronwithdrawing and electron-donating groups at the C6-Ar of purine lead to subtle decrease in reaction yields ([Scheme 3,](#page-2-0) 3gk−3ik). Notably, some strong electron-withdrawing groups such as $-F$, $-CF_3$, and $-CN$ were well tolerated in this photoreaction that is not feasible in the previous ways $⁶$ $⁶$ $⁶$ </sup> [\(Scheme 3,](#page-2-0) 3gh−3gj). Moreover, a range of functional groups $(-Cl, -Br, -CO₂Et)$ could successfully remain in the products, which were useful for the further modifications ([Scheme 3,](#page-2-0) 3gd−3gg and 3gk−3kk). Therefore, 2k was selected as the cross partner to react with challenge purine nucleosides. Deoxyribosyl purine nucleoside and arabinoribosyl purine nucleoside show well anti-HIV/HBV activity, 14 and its

Table 1. Optimization of the Reaction Conditions^a

a Reaction conditions: 1a (0.2 mmol), 2a (1.2 mmol), catalyst (5 mol %), and photoredox catalyst (2.5 mol %) in MeOH (2 mL) at room temperature irradiation with 4 \times 7 W blue LED bulbs under N₂. Isolated yield. Without N_2 protection. d Irradiation with 4×7 W white LED bulbs. e N.R.= No Reaction. f In the absence of light.

Scheme 2. Direct C−H Monophenylation of Various Purines
(Nucleoside) with ArN.RF.^a (Nucleoside) with $ArN₂BF₄$ ^{*}

a Unless otherwise mentioned, all of the reactions were carried out with $1a-1e$ (0.2 mmol), $2a$ (1.2 mmol), Pd(OAc), (5 mol %), $Ru(bpy)_{3}Cl_{2}·6H_{2}O$ (2.5 mol %) in MeOH (2 mL) at room temperature irradiation with 4×7 W blue LED bulbs for 4 h under $\mathbf{N}_{2}.$ b^2 2a (1.6 mmol), 2 h.

derivatives (1j and 1k) proceeded well and affored the corresponding products with good yields at 0 °C (Scheme 3, 3jk and 3kk).

a Reaction conditions: 1g−1i (0.2 mmol), 2b−2k (1.6 mmol), $Pd(OAc)$ ₂ (5 mol %), Ru(bpy)₃Cl₂·6H₂O (2.5 mol %) in MeOH (2 mL) at room temperature irradiation with 4×7 W blue LED bulbs for 2 h under N_2 . \overline{R} At 0 °C.

To further evaluate the prospect of the methodology in synthesis, a gram-scale synthesis of nucleoside analogue 3jk was performed. As shown in [Scheme 4,](#page-3-0) 3jk was obtained in 65% yield (1.06 g) by treatment of 3 mmol of deoxyribosyl purine nucleoside 1j in the presence of 5 mol % $Pd(OAc)₂$, 2.5 mol % of $Ru(bpy)_{3}Cl_{2}·6H_{2}O$, and p-ethoxycarbonylbenzenediazonium tetrafluoroborate 2k. After that, in the present of $NH₃/MeOH$ solution, the deprotection of the product 3jk could be carried out, affording the biaryl-purine nucleoside with free hydroxyl group 4 in 95% yield which can be directly used as an active fragment in a biological test.

On the basis of the previous work and our experimental results,[7a](#page-6-0),[13](#page-7-0) a plausible mechanism is outlined in [Scheme 5](#page-3-0)a: (i) photoexcitation of Ru^{2+} generated $Ru^{2+\ast}$, and aryl radical was formed by single electron transfer from the $Ru^{2+\frac{1}{2}}$; (ii) purinedirected abstraction of 1 with $Pd(OAc)_2$ gave the N1 atommediated palladacycle intermediate A (see [Supporting In](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.7b00659/suppl_file/jo7b00659_si_001.pdf)[formation](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.7b00659/suppl_file/jo7b00659_si_001.pdf) for more details on the preparation of intermediate A); (iii) addition of aryl radical to intermediate A afforded the Pd^{III} intermediate B; (iv) one-electron oxidation of B by $Ru³⁺$ regenerated the photocatalyst and formed Pd^V intermediate C ;

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Scheme 4. Gram-Scaled Synthesis of 3jk and Deprotection of 3jk

condition a: Pd(OAc)₂ (5 mol %), Ru(bpy)₃Cl₂ 6H₂O (2:5 mol %), N₂, 15 mL MeOH, 0 °C, 2 h, 4*7W blue LED condition b: NH₃/MeOH, rt, 1 h

Scheme 5. Plausible Mechanism for Monophenylation Reaction

and (v) the following step involved aryl−aryl bond formation and the regeneration of $Pd(OAc)$ ₂ via reductive elimination. In order to trap the aryl radical generated in the reaction, TEMPO was added to the reaction mixture. Only trace amounts of the target product 1jk was detected, and the radical-trapping product 5 was isolated in 85% yield (Scheme 5b).

In summary, a combined photoredox- and palladiumcatalyzed system for monoselective C_{Ar}−H bond arylation of purine nucleoside has been reported for the first time. The reaction operates at room temperature in only 2−4 h and is mediated by visible light from household blue LED bulbs. Catalytic amounts of photoredox catalyst can generate an aryl radical and enable a $Pd(II)-Pd(III)-Pd(IV)-Pd(II)$ catalytic

cycle, so that external oxidants can be avoided. This C−H arylation approach provided a broad substrate scope reaction condition to a variety of functionalized purines (nucleosides) which are potentially of great importance in medicinal chemistry.

EXPERIMENTAL SECTION

General Information. ¹H NMR spectra were recorded on commercial instruments (400/600 MHz). Chemical shifts are recorded in ppm relative to tetramethylsilane and with the solvent resonance as the internal standard. Data are reported as follows: chemical shift, multiplicity (s = singlet, $d =$ doublet, t = triplet, q = quaternary, $m =$ multiplet, $br =$ broad), coupling constants (Hz), and

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integration. 13C NMR data were collected on commercial instruments (100/150 MHz) with complete proton decoupling. Chemical shifts are reported in ppm from the tetramethylsilane with the solvent resonance as an internal standard. High-resolution mass spectra were taken with electrospray ionization (ESI) as the ionization method used for the HRMS measurement. For column chromatography silica gel (200− 300 mesh) was used as the stationary phase. All regents and solvents were purchased from commercial sources and purified commonly before used. Experiments upon visible-light irradiation were carried out using household blue LED lamps ($\lambda_{\text{max}} = 465 \text{ nm}, 4 \times 7 \text{ W}$).

Synthesis of Starting Materials. Synthesis of 6-Aryl-9- Substituted Purine 1a−1k. Staring materials 1a−1k were synthesized by Suzuki cross-coupling reactions of 6-chloro-9-substituted purine with aryl boronic acid according to the corresponding ref [15](#page-7-0). 6- Chloro-9-substituted purine (3 mmol), boronic acid (4.5 mmol), $K₂CO₃$ (6 mmol), and toluene (20 mL) were added in a 100 mL tube. The tube was refluxed in a 110 °C bath and stirred for 8−12 h under N_2 gas. The mixture was then allowed to cool to room temperature. The mixture was diluted with water, then extracted with ethyl acetate. The extracts were combined, washed with brine, and then dried over anhydrous $Na₂SO₄$. The crude material was purified by column chromatography on silica gel ($V_{\rm PE}/V_{\rm EA}$ = 3:1 as eluent) to give the 6aryl-9-substituted purine.

Synthesis of Diazonium Tetrafluoroborate Salts 2a−2k. Aniline (10 mmol) was dissolved in a mixture of 4 mL of distilled water and 3.4 mL of 50% hydrofluoroboric acid. After cooling the reaction mixture to 0 °C, sodium nitrite (0.69 g) dissolved in 1.5 mL of water was added dropwise. The resulting mixture was stirred for 1 h, and the precipitate was collected by filtration, washed with water, and redissolved in a minimum amount of acetone. Diethyl ether was added until precipitation of the corresponding diazonium tetrafluoroborate, which was filtered, washed several times with diethyl ether, and dried under vacuum.

General Procedure for the Monoselective Ortho C−H Arylation. In a 25 mL sealed tube was charged with 1a−1e (0.2 mmol), Pd(OAc)₂ (2.2 mg, 0.01 mmol), Ru(bpy)₃Cl₂·6H₂O (3.7 mg, 0.005 mmol), and phenyldiazonium tetrafluoroborate 2a (1.2 mmol), then MeOH (2 mL) were added, and the reaction mixture was irradiated with 4×7 W blue LED bulbs for 4 h at room temperature under a N_2 atmosphere via several vacuum- N_2 exchanges. A 25 mL sealed tube was charged with 1f−1i (0.2 mmol), Pd(OAc)₂ (2.2 mg, 0.01 mmol), $Ru(bpy)_{3}Cl_{2}.6H_{2}O$ (3.7 mg, 0.005 mmol), and phenyldiazonium tetrafluoroborate 2a−2k (1.6 mmol), then MeOH (2 mL) was added, and the reaction mixture was irradiated with 4×7 W blue LED bulbs for 2 h at room temperature under a N_2 atmosphere via several vacuum- N_2 exchanges. A 25 mL sealed tube was charged with 1j−1k (0.2 mmol), Pd(OAc)₂ (2.2 mg, 0.01 mmol), $Ru(bpy)_{3}Cl_{2}\cdot 6H_{2}O$ (3.7 mg, 0.005 mmol), and phenyldiazonium tetrafluoroborate 2k (1.6 mmol), then MeOH (2 mL) was added, and the reaction mixture was irradiated with 4×7 W blue LED bulb for 2 h at 0 °C under a N_2 atmosphere via several vacuum- N_2 exchanges. After completion of the reaction, the mixture was then diluted with brine (10 mL) and extracted with ethyl acetate (3×10 mL). The organic layer was collected and dried over anhydrous $Na₂SO₄$, filtered, and concentrated under reduced pressure. The crude material was purified by column chromatography on silica gel $(V_{PE}/V_{EA} = 1:1$ as eluent) to give the desired products 3aa−3kk.

Deprotection of Diacetyl Deoxypurine Nucleoside Ana**logues.** In the presence of $NH₃/MeOH$ solution (7 mol/L), the deprotection of the product 3jk (81 mg, 0.15 mmol) could be carried out, affording the biaryl-purine nucleoside with free hydroxyl group 4 (66 mg) in 95% yield.

The Background Reaction with TEMPO. A 25 mL sealed tube was charged with 1j (39.6 mg, 0.1 mmol), $Pd(OAc)_{2}$ (1.1 mg, 0.005 mmol), $Ru(bpy)$ ₃Cl₂·6H₂O (1.3 mg, 0.0025 mmol), phenyldiazonium tetrafluoroborate 2k (158.4 mg, 0.6 mmol), and TEMPO (93.8 mg, 0.6 mmol), then MeOH (2 mL) was added, and the reaction mixture was irradiation with 4×7 W blue LED bulbs for 4 h at 0 $^{\circ}$ C under a N_2 atmosphere via several vacuum- N_2 exchanges. After completion of the reaction, the mixture was then diluted with brine (10 mL) and

extracted with ethyl acetate $(3 \times 10 \text{ mL})$. The organic layer was collected, dried over anhydrous $Na₂SO₄$, filtered, and concentrated under reduced pressure. The crude material was purified by column chromatography on silica gel ($V_{PE}/V_{EA} = 10:1$ as eluent) to give the desired products 5 (155 mg, 85%).

Characterization of Compounds. 6-([1,1′-Biphenyl]-2-yl)-9 benzyl-9H-purine (3aa). Light-yellow oil. Yield 53 mg (73%) at 0.2 mmol scale. ¹H NMR (400 MHz, CDCl₃) δ 8.89 (s, 1H), 7.87 (s, 1H), 7.80 (d, J = 8.0 Hz, 1H), 7.57−7.49 (m, 3H), 7.39−7.34 (m, 3H), 7.22−7.20 (dd, J = 8.0 Hz, 2 Hz, 2H), 7.15−7.11 (m, 5H), 5.41 (s, 2H); ¹³C NMR (151 MHz, CDCl₃) δ 159.1, 152.5, 151.6, 144.2, 142.0, 141.3, 135.2, 134.4, 132.2, 130.9, 130.7, 129.9, 129.2, 129.1, 128.6, 127.8, 127.6, 127.4, 126.5, 47.2. HRMS (ESI-TOF) m/z: [M + $[H]^+$ calcd for $C_{24}H_{19}N_4$ 363.1604; found 363.1609.

6-([1,1′-Biphenyl]-2-yl)-9-methyl-9H-purine (3ba). Light-yellow oil. Yield 49 mg (86%) at 0.2 mmol scale. ¹H NMR (400 MHz, CDCl₃) δ 8.86 (s, 1H), 7.91 (s, 1H), 7.78 (d, J = 7.2 Hz, 1H), 7.57– 7.50 (m, 3H), 7.16−7.12 (m, 5H), 3.87 (s, 3H); 13C NMR (151 MHz, CDCl3) δ 157.9, 151.2, 150.8, 143.9, 140.9, 140.3, 133.3, 131.2, 130.0, 129.8, 128.8, 128.2, 126.8, 126.3, 125.5, 28.8. HRMS (ESI-TOF) m/z: $[M + Na]$ ⁺ calcd for $C_{18}H_{14}N_4Na$ 309.1111; found 309.1101.

6-([1,1′-Biphenyl]-2-yl)-9-(sec-butyl)-9H-purine (3ca). Brown oil. Yield 54 mg (83%) at 0.2 mmol scale. ¹H NMR (400 MHz, CDCl₃) δ 8.83 (s, 1H), 7.91 (s, 1H), 7.80 (d, J = 7.6 Hz, 1H), 7.56−7.49 (m, 3H), 7.15−7.09 (m, 5H), 4.64 (m, 1H), 2.05−1.88 (m, 2H), 1.59 (d, J $= 6.8$ Hz, 3H), 0.80 (t, J = 7.4 Hz, 3H); ¹³C NMR (151 MHz, CDCl₃) δ 158.9, 151.9, 151.4, 142.6, 141.9, 141.3, 134.5, 132.5, 130.9, 130.7, 129.8, 129.2, 127.8, 127.3, 126.4, 53.0, 29.5, 20.4, 10.6. HRMS (ESI-TOF) m/z : $[M + Na]^+$ calcd for $C_{21}H_{20}N_4N_4$ 351.1580; found 351.1579.

6-([1,1′-Biphenyl]-2-yl)-9-cyclopentyl-9H-purine (3da). Brown oil. Yield 53 mg (78%) at 0.2 mmol scale. NMR (600 MHz,) δ 8.82 (s, 1H), 7.97 (s, 1H), 7.79 (d, J = 7.5 Hz, 1H), 7.55−7.49 (m, 3.0 Hz), 7.16−7.12 (m, 5H), 4.97 (m, 1H), 2.32−2.29(m, 2H), 2.04−1.99 (m, 2H), 1.95−1.93 (m, 2H), 1.83−1.81 (m, 2H).; 13C NMR (151 MHz, CDCl3) δ 157.9, 150.8, 150.6, 141.6, 140.8, 140.3, 133.4, 131.7, 130.0, 129.8, 128.7, 128.2, 126.8, 126.3, 125.4, 55.1, 31.6, 22.9. HRMS (ESI-TOF) m/z : $[M + H]^+$ calcd for $C_{22}H_{21}N_4$ 341.1761; found 341.1756.

6-([1,1′-Biphenyl]-2-yl)-9-(2-bromoethyl)-9H-purine (3ea). Lightbrown oil. Yield 56 mg (74%) at 0.2 mmol scale. $^1\rm H$ NMR (400 MHz, CDCl₃) δ 8.84 (s, 1H), 7.99 (s, 1H), 7.82 (d, J = 7.2 Hz, 1H), 7.58– 7.52 (m, 3H), 7.16−7.12 (m, 5H), 4.65 (t, J = 5.8 Hz, 2H), 3.79 (t, J = 6.0 Hz, 2H); ¹³C NMR (151 MHz, CDCl₃) δ 159.3, 152.3, 151.3, 144.5, 142.0, 141.2, 134.2, 132.4, 131.0, 130.8, 129.9, 129.2, 127.9, 127.4, 126.5, 45.6, 29.6. HRMS (ESI-TOF) m/z: [M + Na]⁺ calcd for $C_{19}H_{15}BrN_4Na$ 401.0372; found 401.0376.

Ethyl 2-(6-([1,1'-Biphenyl]-2-yl)-9H-purin-9-yl)acetate (3fa). Yellow oil. Yield 49 mg (68%) at 0.2 mmol scale. ¹H NMR (400 MHz, CDCl₃) δ 8.84 (s, 1H), 7.99 (s, 1H), 7.82 (d, J = 7.2 Hz, 1H), 7.58– 7.50 (m, 3H), 7.17−7.13 (m, 5H), 5.01 (s, 2H), 4.27 (q, J = 7.2 Hz, 2H), 1.30 (t, J = 7.1 Hz, 3H). ¹³C NMR (151 MHz, CDCl₃) δ 166.9, 159.2, 152.4, 151.6, 144.6, 141.9, 141.2, 134.2, 131.8, 131.1, 130.8, 129.9, 129.2, 127.9, 127.4, 126.5, 62.5, 44.1, 14.1. HRMS (ESI-TOF) m/z : [M + Na]⁺ calcd for C₂₁H₁₈N₄O₂Na 381.1322; found 381.1322.

((3aR,4R,6R,6aR)-6-(6-([1,1′-Biphenyl]-2-yl)-9H-purin-9-yl)-2,2 dimethyltetrahydrofuro[3,4-d][1,3]dioxol-4-yl)methanol (3ga). Light-rellow oil. Yield 56 mg $(63%)$ at 0.2 mmol scale. ${}^{1}H$ NMR $(400 \text{ MHz}, \text{CDCl}_3)$ δ 8.79 (s, 1H), 8.02 (s, 1H), 7.78 (d, J = 7.6 Hz, 1H), 7.58−7.50 (m, 3H), 7.16 (s, 5H), 5.91 (d, J = 4.8 Hz, 1H), 5.76 $(d, J = 9.6, 1H)$, 5.24 $(t, J = 5.2$ Hz, 1H), 5.11 $(d, J = 5.2, 1H)$, 4.54 $(s,$ 1H), 3.96 (d, J = 12.4 Hz, 1H), 3.78 (t, J = 11.0 Hz, 1H), 1.65 (s, 3H), 1.38 (s, 3H); ¹³C NMR (151 MHz, CDCl₃) δ 160.4, 151.6, 150.2, 144.1, 142.0, 141.0, 133.9, 133.8, 131.2, 131.0, 130.1, 129.2, 128.0, 127.4, 126.7, 114.3, 94.1, 86.1, 82.9, 81.6, 63.3, 27.6, 25.3. HRMS (ESI-TOF) m/z : $[M + H]^+$ calcd for $C_{25}H_{25}N_4O_4$ 445.1870; found 445.1875.

((3aR,4R,6R,6aR)-2,2-Dimethyl-6-(6-(4′-methyl-[1,1′-biphenyl]-2 yl)-9H-purin-9-yl)tetrahydrofuro[3,4-d][1,3]dioxol-4-yl)methanol (3gb). Light-brown oil. Yield 62 mg (68%) at 0.2 mmol scale. $^1\mathrm{H}$ NMR (600 MHz, CDCl₃) δ 8.80 (s, 1H), 8.01 (s, 1H), 7.74 (d, J = 7.5 Hz, 1H), 7.57–7.53 (m, 2H), 7.49 (t, J = 7.2 Hz, 1H), 7.03 (d, J = 7.8 Hz, 2H), 6.97 (d, J = 7.8 Hz, 2H), 5.91 (d, J = 4.8 Hz, 1H), 5.79 (d, J $= 10.8$ Hz, 1H), 5.25 (t, J = 5.1 Hz, 1H), 5.13 (d, J = 6.0 Hz, 1H), 4.56 $(s, 1H)$, 3.99 (d, J = 12.6 Hz, 1H), 3.81 (t, J = 12.0 Hz, 1H), 2.28 (s, 3H), 1.66 (s, 3H), 1.39 (s, 3H); ¹³C NMR (151 MHz, CDCl₃) δ 160.8, 151.7, 150.3, 144.2, 142.1, 138.2, 136.5, 134.2, 133.9, 131.3, 131.1, 130.3, 129.2, 128.9, 127.3, 114.4, 94.4, 86.2, 83.0, 81.8, 63.5, 27.8, 25.4, 21.2. HRMS (ESI-TOF) m/z : $[M + Na]^{+}$ calcd for $C_{26}H_{26}N_4NaO_4$ 481.1846; found 481.1851.

((3aR,4R,6R,6aR)-6-(6-(4′-Methoxy-[1,1′-biphenyl]-2-yl)-9Hpurin-9-yl)-2,2-dimethyltetrahydrofuro[3,4-d][1,3]dioxol-4-yl) methanol (3gc). Light-yellow oil. Yield 62 mg (65%) at 0.2 mmol scale. ¹H NMR (400 MHz, CDCl₃) δ 8.82 (s, 1H), 8.00 (s, 1H), 7.73 $(d, J = 7.2 \text{ Hz}, 1H), 7.57–7.46 \text{ (m, 3H)}, 7.07 \text{ (d, } J = 8.8 \text{ Hz}, 2H), 6.70 \text{ }$ $(d, J = 8.8 \text{ Hz}, 2H), 5.90 \ (d, J = 4.8 \text{ Hz}, 1H), 5.76 \ (d, J = 10.0 \text{ Hz},$ 1H), 5.25 (t, $J = 5.4$ Hz, 1H), 5.13 (d, $J = 5.6$ Hz, 1H), 4.56 (s, 1H), 3.99 (d, $J = 12.4$ Hz, 1H), 3.81 (t, $J = 11.2$ Hz, 1H), 3.75 (s, 3H), 1.66 (s, 3H), 1.39 (s, 3H). ¹³C NMR (151 MHz, CDCl₃) δ 159.7, 157.5, 150.6, 149.1, 143.0, 140.6, 133.0, 132.7, 132.5, 130.1, 129.8, 129.3, 129.1, 126.0, 113.3, 112.5, 93.2, 85.1, 81.8, 80.6, 62.3, 54.1, 26.6, 24.2. HRMS (ESI-TOF) m/z : $[M + Na]^+$ calcd for $C_{26}H_{26}N_4NaO_5$ 497.1795; found 497.1795.

((3aR,4R,6R,6aR)-6-(6-(4′-Chloro-[1,1′-biphenyl]-2-yl)-9H-purin-9-yl)-2,2-dimethyltetrahydrofuro[3,4-d][1,3]dioxol-4-yl)methanol (3gd). Brown oil. Yield 68 mg (71%) at 0.2 mmol scale. $^1\text{H NMR}$ (600 MHz, CDCl₃) δ 8.81 (s, 1H), 8.02 (s, 1H), 7.80–7.78 (m, 1H), 7.60– 7.50 (m, 3H), 7.15 (d, $I = 8.4$ Hz, 2H), 7.08 (d, $I = 8.4$ Hz, 2H), 5.92 $(d, J = 4.8 \text{ Hz}, 1H), 5.66 (d, J = 10.4 \text{ Hz}, 1H), 5.26 (t, J = 5.4 \text{ Hz}, 1H),$ 5.13 (d, J = 6.0 Hz, 1H), 4.56 (s, 1H), 3.99 (d, J = 12.8 Hz, 1H), 3.81 $(d, J = 11.2 \text{ Hz}, 1H)$, 1.66 (s, 3H), 1.39 (s, 3H). ¹³C NMR (151 MHz, CDCl3) δ 158.9, 150.6, 149.2, 143.2, 139.8, 138.6, 132.8, 132.7, 131.8, 130.4, 129.8, 129.5, 129.2, 129.0, 127.7, 127.2, 126.7, 113.3, 93.1, 85.1, 81.9, 80.6, 62.3, 26.6, 24.2. HRMS (ESI-TOF) m/z: [M + H]+ calcd for $C_{25}H_{24}CIN_4O_4$ 479.1481; found 479.1481.

((3aR,4R,6R,6aR)-6-(6-(2′-Bromo-[1,1′-biphenyl]-2-yl)-9H-purin-9-yl)-2,2-dimethyltetrahydrofuro[3,4-d][1,3]dioxol-4-yl)methanol (3ge). Light-brown oil. Yield 55 mg $(53%)$ at 0.2 mmol scale. ${}^{1}H$ NMR (400 MHz, CDCl₃) δ 8.63 (s, 1H), 8.07 (s, 1H), 8.05−8.01 (m, 1H), 7.60−7.56 (m, 2H), 7.46−7.42 (m, 2H), 7.26−7.23 (m, 1H), 7.19 (t, $J = 7.6$ Hz, 1H), 7.07 (t, $J = 7.2$ Hz, 1H), 5.93 (d, $J = 4.9$ Hz, 1H), 5.68 (t, J = 11.2 Hz, 1H), 5.24 (q, J = 5.2 Hz, 1H), 5.13–5.10 (m, 1H), 4.55 (s, 1H), 4.00−3.91 (m, 1H), 3.79 (t, J = 11.2 Hz, 1H), 1.66 (s, 3H), 1.39 (s, 3H). ¹³C NMR (151 MHz, CDCl₃) δ 159.1, 151.1, 150.3, 144.1, 142.0, 141.1, 132.3, 131.8, 131.5, 129.9, 128.4, 128.0, 126.8, 114.3, 94.3, 94.2, 81.6, 63.4, 27.7, 25.2. HRMS (ESI-TOF) m/z: $[M + Na]^{+}$ calcd for $C_{25}H_{23}B_{1}NaO_{4}$ 545.0795; found 545.0791.

((3aR,4R,6R,6aR)-6-(6-(3′-Bromo-[1,1′-biphenyl]-2-yl)-9H-purin-9-yl)-2,2-dimethyltetrahydrofuro[3,4-d][1,3]dioxol-4-yl)methanol (3gf). Brown oil. Yield 73 mg (70%) at 0.2 mmol scale. $^1\rm H$ NMR (400 MHz, CDCl₃) δ 8.82 (s, 1H), 8.02 (s, 1H), 7.82 (d, J = 7.2 Hz, 1H), 7.57−7.53 (m, 3H), 7.27 (d, J = 7.2 Hz, 1H), 7.04−6.94 (m, 2H), 5.92 $(d, J = 4.8 \text{ Hz}, 1H), 5.56 (d, J = 10.4 \text{ Hz}, 1H), 5.25 (t, J = 5.2 \text{ Hz}, 1H),$ 5.13 (d, $J = 6.0$ Hz, 1H), 4.55 (s, 1H), 3.98 (d, $J = 12.4$ Hz, 1H), 3.81 $(t, J = 11.4 \text{ Hz}, 1\text{H}), 1.66 \text{ (s, 3H)}, 1.39 \text{ (s, 3H)}.$ ¹³C NMR (101 MHz, CDCl₃) δ 159.9, 151.7, 150.3, 144.3, 140.8, 140.1, 133.7, 131.4, 131.2, 130.8, 130.8, 130.3, 128.7, 128.3, 127.7, 126.5, 121.1, 114.3, 94.2, 86.1, 82.9, 81.6, 63.4, 27.6, 25.2. HRMS (ESI-TOF) m/z : [M + H]⁺ calcd for C25H24BrN4O4 523.0975; found 523.0979.

((3aR,4R,6R,6aR)-6-(6-(4′-Bromo-[1,1′-biphenyl]-2-yl)-9H-purin-9-yl)-2,2-dimethyltetrahydrofuro[3,4-d][1,3]dioxol-4-yl)methanol (3gg). Light-brown oil. Yield 76 mg $(73%)$ at 0.2 mmol scale. 1 H NMR (600 MHz, CDCl₃) δ 8.81 (s, 1H), 8.02 (s, 1H), 7.79 (d, J = 7.5 Hz, 1H), 7.59–7.50 (m, 3H), 7.30 (d, J = 7.8 Hz, 2H), 7.03 (d, J = 7.8 Hz, 2H), 5.92 (d, J = 4.8 Hz, 1H), 5.54 (d, J = 10.8 Hz, 1H), 5.26 (t, J $= 5.4$ Hz, 1H), 5.13 (d, J = 6.0 Hz, 1H), 4.56 (s, 1H), 3.99 (d, J = 12.6 Hz, 1H), 3.81 (t, J = 11.4 Hz, 1H), 1.66 (s, 3H), 1.39 (s, 3H). ¹³C NMR (151 MHz, CDCl₃) δ 159.9, 151.7, 150.3, 140.8, 140.1, 133.9, 133.7, 131.4, 131.2, 130.8, 130.8, 130.3, 127.8, 121.1, 114.3, 94.2, 86.1, 82.9, 63.4, 27.6, 25.2. HRMS (ESI-TOF) m/z : $[M + H]^+$ calcd for $C_{25}H_{24}B_{r}N_{4}O_{4}$ 523.0975; found 523.0973.

((3aR,4R,6R,6aR)-6-(6-(4′-Fluoro-[1,1′-biphenyl]-2-yl)-9H-purin-9 yl)-2,2-dimethyltetrahydrofuro[3,4-d][1,3]dioxol-4-yl)methanol (3gh). Brown oil. Yield 62 mg (67%) at 0.2 mmol scale. $^1\rm H$ NMR (600 MHz, CDCl₃) δ 8.81 (s, 1H), 8.00 (s, 1H), 7.77 (d, J = 7.2 Hz, 1H), 7.57 (t, J = 7.8 Hz, 1H), 7.52 (t, J = 7.2 Hz, 2H), 7.11 (dd, J = 8.2, 5.6 Hz, 2H), 6.86 (t, J = 8.4 Hz, 2H), 5.91 (d, J = 4.8 Hz, 1H), 5.59 (d, J = 10.8 Hz, 1H), 5.25 (t, $J = 5.4$ Hz, 1H), 5.12 (d, $J = 6.0$ Hz, 1H), 4.55 $(s, 1H)$, 3.98 (d, J = 12.6 Hz, 1H), 3.80 (t, J = 11.7 Hz, 1H), 1.65 (s, 3H), 1.39 (s, 3H). ¹³C NMR (151 MHz, CDCl₃) δ 162.7, 161.1, 160.1, 151.6, 150.2, 144.2, 141.0, 137.1, 133.9, 133.8, 131.3, 130.9, 130.8, 130.7, 130.2, 129.2, 128.0, 127.5, 115.0, 114.3, 94.2, 86.1, 82.9, 81.6, 63.3, 27.6, 25.2. HRMS (ESI-TOF) m/z : $[M + H]^{+}$ calcd for $C_{25}H_{24}FN_{4}O_{4}$ 463.1776; found 463.1779.

((3aR,4R,6R,6aR)-2,2-Dimethyl-6-(6-(4′-(trifluoromethyl)-[1,1′-biphenyl]-2-yl)-9H-purin-9-yl)tetrahydrofuro[3,4-d][1,3]dioxol-4-yl) methanol (3gi). Light yellow oil. Yield 71 mg (69%) at 0.2 mmol scale. ¹H NMR (400 MHz, CDCl₃) δ 8.76 (s, 1H), 8.01 (s, 1H), 7.83 $(d, J = 7.2 \text{ Hz}, 1H), 7.61–7.49 \text{ (m, 3H)}, 7.41 \text{ (d, } J = 8.0 \text{ Hz}, 2H), 7.24$ $(s, 2H)$, 5.90 (d, J = 5.2 Hz, 1H), 5.46 (d, J = 10.0 Hz, 1H), 5.22 (t, J = 5.2 Hz, 1H), 5.10 (d, $J = 5.2$ Hz, 1H), 4.53 (s, 1H), 3.97 (d, $J = 12.8$ Hz, 1H), 3.79 (t, $J = 11.2$ Hz, 1H), 1.63 (s, 3H), 1.37 (s, 3H). ¹³C NMR (151 MHz, CDCl₃) δ 159.5, 151.6, 150.4, 144.9, 144.3, 140.7, 133.9, 133.8, 131.6, 131.0, 130.4, 129.5, 128.1, 125.0, 124.9, 114.4, 94.2, 86.1, 82.9, 81.6, 63.3, 27.6, 25.2. HRMS (ESI-TOF) m/z: [M + H ⁺ calcd for $C_{26}H_{24}F_{3}N_{4}O_{4}$ 513.1744; found 513.1747.

2′-(9-((3aR,4R,6R,6aR)-6-(Hydroxymethyl)-2,2-dimethyltetrahydrofuro[3,4-d][1,3]dioxol-4-yl)-9H-purin-6-yl)-[1,1′-biphenyl]-4 carbonitrile (3gj). Yellow oil. Yield 70 mg $(75%)$ at 0.2 mmol scale. ¹H NMR (400 MHz, CDCl₃) δ 8.79 (s, 1H), 8.02 (s, 1H), 7.88 (d, J = 7.6 Hz, 1H), 7.64−7.57 (m, 2H), 7.51 (dd, J = 7.6, 2.0 Hz, 1H), 7.47 $(d, J = 8.4 \text{ Hz}, 2H), 7.27 \text{ (s, 1H)}, 7.25 \text{ (s, 1H)}, 5.92 \text{ (d, } J = 4.8 \text{ Hz},$ 1H), 5.43 (d, J = 9.6 Hz, 1H), 5.24 (t, J = 5.2 Hz, 1H), 5.12 (d, J = 4.8 Hz, 1H), 4.55 (s, 1H), 3.98 (d, $J = 12.8$ Hz, 1H), 3.81 (t, $J = 11.2$ Hz, 1H), 1.66 (s, 3H), 1.39 (s, 3H). ¹³C NMR (151 MHz, CDCl₃) δ 159.1, 151.7, 150.4, 146.1, 144.4, 140.2, 133.8, 131.8, 131.8, 130.8, 130.5, 129.9, 128.5, 118.9, 114.4, 110.5, 94.1, 86.1, 63.33. HRMS (ESI-TOF) m/z : $[M + H]^+$ calcd for $C_{26}H_{24}N_5O_4$ 470.1823; found 470.1826.

Ethyl 2′-(9-((3aR,4R,6R,6aR)-6-(Hydroxymethyl)-2,2-dimethyltetrahydrofuro [3,4-d][1,3]dioxol-4-yl)-9H-purin-6-yl)-[1,1′-biphenyl]- 4-carboxylate (3gk). Light-brown oil. Yield 85 mg $(82%)$ at 0.2 mmol scale. ¹H NMR (400 MHz, CDCl₃) δ 8.78 (s, 1H), 8.01 (s, 1H), 7.86– 7.81 (m, 3H), 7.62−7.53 (m, 3H), 7.22 (d, J = 8.4 Hz, 2H), 5.91 (d, J $= 4.8$ Hz, 1H), 5.61 (d, J = 9.6 Hz, 1H), 5.25 (t, J = 5.4 Hz, 1H), 5.12 $(d, J = 6.0 \text{ Hz}, 1H)$, 4.55 (s, 1H), 4.33 (q, J = 7.2 Hz, 2H), 3.98 (d, J = 12.8 Hz, 1H), 3.80 (t, J = 11.2 Hz, 1H), 1.65 (s, 3H), 1.39–1.35(m, 6H). ¹³C NMR (151 MHz, CDCl₃) δ 166.6, 159.7, 151.6, 150.3, 145.9, 144.2, 141.1, 133.9, 132.1, 132.1, 132.0, 131.4, 130.8, 130.3, 129.3, 129.2, 128.7, 128.6, 128.5, 128.0, 114.3, 94.2, 86.2, 82.9, 63.3, 60.9, 27.6, 25.2, 14.3. HRMS (ESI-TOF) m/z : $[M + Na]^{+}$ calcd for $C_{28}H_{28}N_4NaO_6$ 539.1901; found 539.1900.

Ethyl 2′-(9-((3aR,4R,6R,6aR)-6-(Hydroxymethyl)-2,2-dimethyltebiphenyl]-4-carboxylate (3hk). Brown oil. Yield 76 mg (72%) at 0.2 mmol scale. ¹H NMR (400 MHz, CDCl₃) δ 8.75 (s, 1H), 7.98 (s, 1H), 7.84 (d, J = 8.4 Hz, 2H), 7.75 (d, J = 8.4 Hz, 1H), 7.37 (d, J = 6.8 Hz, 2H), 7.22 (d, J = 8.4 Hz, 2H), 5.90 (d, J = 4.8 Hz, 1H), 5.58 (d, J = 10.8 Hz, 1H), 5.25 (t, $J = 5.4$ Hz, 1H), 5.12 (d, $J = 6.0$ Hz, 1H), 4.54 $(s, 1H)$, 4.34 $(q, J = 7.2$ Hz, 2H), 3.97 $(d, J = 12.8$ Hz, 1H), 3.80 $(t, J = 1.2)$ 11.4 Hz, 1H), 2.48 (s, 3H), 1.65 (s, 3H), 1.39−1.35 (m, 6H). 13C NMR (151 MHz, CDCl₃) δ 166.6, 159.8, 151.6, 150.2, 146.1, 144.0, 141.0, 140.5, 133.9, 131.6, 131.5, 131.1, 129.2, 129.2, 128.8, 128.6, 114.3, 94.2, 86.1, 82.8, 81.6, 63.4, 60.9, 27.6, 25.2, 21.5, 14.3. HRMS (ESI-TOF) m/z : [M + H]⁺ calcd for C₂₉H₃₁N₄O₆ 531.2238; found 531.2237.

Diethyl 6-(9-((3aR,4R,6R,6aR)-6-(Hydroxymethyl)-2,2-dimethyltetrahydrofuro [3,4-d][1,3]dioxol-4-yl)-9H-purin-6-yl)-[1,1′-biphenyl]- 3,4'-dicarboxylate (3ik). Light-yellow oil. Yield 74 mg $(63%)$ at 0.2 mmol scale. ¹H NMR (600 MHz, CDCl₃) δ 8.81 (s, 1H), 8.22–8.21 $(s, 2H)$, 8.01 $(s, 1H)$, 7.90 $(d, J = 7.8 \text{ Hz}, 1H)$, 7.87 $(d, J = 7.8 \text{ Hz},$ 2H), 7.25 (d, J = 7.8 Hz, 2H), 5.91 (d, J = 4.8 Hz, 1H), 5.36 (d, J =

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11.4 Hz, 1H), 5.24 (t, $J = 5.1$ Hz, 1H), 5.12 (d, $J = 6.0$ Hz, 1H), 4.55 $(s, 1H)$, 4.44 $(q, J = 7.2$ Hz, 2H), 4.35 $(q, J = 7.2$ Hz, 2H), 3.97 $(d, J =$ 12.6 Hz, 1H), 3.80 (t, $J = 11.7$ Hz, 1H), 1.65 (s, 3H), 1.43 (t, $J = 7.2$ Hz, 3H), 1.39–1.36 (m, 6H). ¹³C NMR (151 MHz, CDCl₃) δ 166.4, 165.9, 158.6, 151.7, 150.4, 144.9, 144.5, 141.4, 137.9, 133.9, 129.4, 129.2, 114.4, 94.1, 86.2, 82.9, 81.6, 63.3, 61.4, 61.0, 30.9, 27.6, 25.2, 14.3. HRMS (ESI-TOF) m/z : $[M + H]^+$ calcd for $C_{31}H_{33}N_4O_8589.2293$; found 589.2295.

 $EthyL$ $2'$ -(9-((2R,4S,5R)-4-Acetoxy-5-(acetoxymethyl)tetrahydrofuran-2-yl)-9H-purin-6-yl)-[1,1′-biphenyl]-4-carboxylate (3jk). Colorless oil. Yield 85 mg $(78%)$ at 0.2 mmol scale. ^{1}H NMR (600 MHz, CDCl₃) δ 8.82 (s, 1H), 8.11 (s, 1H), 7.83 (t, J = 7.8 Hz, 3H), 7.60−7.54 (m, 3H), 7.23 (d, J = 7.8 Hz, 2H), 6.48 (t, J = 7.2 Hz, 1H), 5.44 (d, J = 6.0 Hz, 1H), 4.40 (q, J = 6.0 Hz, 1H), 4.36−4.31 (m, 4H), 3.00−2.95 (m, 1H), 2.67−2.63 (m, 1H), 2.14 (s, 3H), 2.06 (s, 3H), 1.36 (t, J = 7.2 Hz, 3H). ¹³C NMR (151 MHz, CDCl₃) δ 170.4, 170.3, 166.6, 158.8, 152.3, 151.1, 146.0, 142.6, 140.9, 134.2, 132.9, 131.3, 130.8, 130.1, 129.2, 129.2, 128.6, 128.0, 84.7, 82.7, 74.5, 63.7, 60.9, 37.5, 20.9, 20.7, 14.3. HRMS (ESI-TOF) m/z: [M + Na]⁺ calcd

for $C_{29}H_{28}N_4N_4O_7$ 567.1850; found 567.1849.
(2R,3R,4S,5R)-2-(Acetoxymethyl)-5-(6-(4'-(ethoxycarbonyl)-[1,1'biphenyl]-2-yl)-9H-purin-9-yl)tetrahydrofuran-3,4-diyl diacetate $(3kk)$. Colorless oil. Yield 98 mg $(81%)$ at 0.2 mmol scale. ¹H NMR (400 MHz, CDCl₃) δ 8.87 (s, 1H), 8.08 (s, 1H), 7.83-7.79 (m, 3H), 7.60−7.53 (m, 3H), 7.23 (s, 1H), 7.21 (s, 1H), 6.62 (d, J = 4.4 Hz, 1H), 5.48 (t, J = 3.8 Hz, 1H), 5.43 (t, J = 3.6 Hz 1H), 4.45 (t, J = 4.8 Hz, 2H), 4.31 (q, J = 7.2 Hz, 2H), 4.26 (m, 1H), 2.17 (s, 3H), 2.10 $(s, 3H)$, 1.78 $(s, 3H)$, 1.34 $(t, J = 7.2 \text{ Hz}, 3H)$. ¹³C NMR (151 MHz, CDCl3) δ 170.5, 169.6, 168.5, 166.5, 158.7, 152.6, 150.9, 146.0, 143.3, 140.9, 134.1, 131.9, 129.2, 128.6, 128.1, 83.1, 80.0, 75.8, 74.9, 62.9, 60.9, 20.8, 20.7, 20.1, 14.3. HRMS (ESI-TOF) m/z: [M + Na]⁺ calcd for $C_{31}H_{30}N_4NaO_9$ 625.1905; found 625.1903.

 $Ethy\tilde{\nu}$ 2'-(9-((2R,4S,5R)-4-Hydroxy-5-(hydroxymethyl)tetrahydrofuran-2-yl)-9H-purin-6-yl)-[1,1′-biphenyl]-4-carboxylate (4). White solid. Yield 68 mg (95%) at 0.2 mmol scale of 3j. Mp 62− 64 °C. ¹H NMR (600 MHz, CDCl₃) δ 8.78 (s, 1H), 8.00 (s, 1H),7.84 $(m, 3H)$, 7.61–7.54 (t, J = 8.4 Hz, 3H), 7.23 (d, J = 7.8 Hz, 2H), 6.36 $(q, J = 5.2 \text{ Hz}, 1\text{H})$, 5.62 (s, 1H), 4.80 (d, J = 4.8 Hz, 1H), 4.34 (q, J = 7.2 Hz, 2H), 4.23 (s, 1H), 3.98 (dd, J = 13.2, 5.4 Hz, 1H), 3.88 (s, 1H), 3.80 (d, J = 11.4 Hz, 1H), 3.12−3.08 (m, 1H), 2.35 (q, J = 5.4 Hz, 1H), 1.37 (t, J = 7.2 Hz, 3H). ¹³C NMR (151 MHz, CDCl₃) δ 166.6, 159.6, 151.4, 150.4, 145.9, 144.3, 141.1, 134.1, 134.0, 131.4, 130.8, 130.3, 129.3, 129.2, 128.7, 128.0, 89.6, 87.7, 73.5, 63.4, 61.0, 40.7, 14.3. HRMS (ESI-TOF) m/z : $[M + Na]$ ⁺ calcd for $C_{25}H_{24}N_{4}NaO_{5}$ 483.1639; found 483.1632.

Ethyl 4-((2,2,6,6-Tetramethylpiperidin-1-yl)oxy)benzoate (5). Browm solid. Yield 155 mg (85%) at 0.6 mmol scale.¹H NMR (600 MHz, CDCl₃) δ 7.94 (d, J = 9.0 Hz, 2H), 7.20 (br, 2H), 4.33 (q, J = 7.2 Hz, 2H), 4.06 (br, 1H), 1.94−1.91 (m, 3H), 1.63 (t, J = 12.0 Hz, 2H), 1.36 (t, J = 7.2 Hz, 3H), 1.27 (s, 6H), 1.01 (s, 6H). 13C NMR (101 MHz, CDCl3) δ 167.1, 166.6, 131.1, 122.6, 113.7, 62.9, 60.9, 60.6, 48.2, 32.5, 21.3, 14.4. HRMS (ESI-TOF) m/z: [M + H]+ calcd for $C_{18}H_{28}NO_3$ 306.2064; found 306.2067.

■ ASSOCIATED CONTENT

S Supporting Information

The Supporting Information is available free of charge on the [ACS Publications website](http://pubs.acs.org) at DOI: [10.1021/acs.joc.7b00659.](http://pubs.acs.org/doi/abs/10.1021/acs.joc.7b00659)

Verification test, copies of all spectral and full characterization for all new compounds [\(PDF](http://pubs.acs.org/doi/suppl/10.1021/acs.joc.7b00659/suppl_file/jo7b00659_si_001.pdf))

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Notes

The authors declare no competing financial interest.

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